

**Subject: GCSE Chemistry Triple Science****Year: 10**

<b><u>Autumn HT1</u></b> <b><u>(approx. 21</u></b> <b><u>lessons)</u></b>	<b><u>Autumn HT2</u></b> <b><u>(approx. 21</u></b> <b><u>lessons)</u></b>	<b><u>Spring HT1</u></b> <b><u>(approx. 15</u></b> <b><u>lessons)</u></b>	<b><u>Spring HT2</u></b> <b><u>(approx. 12</u></b> <b><u>lessons)</u></b>	<b><u>Summer HT1</u></b> <b><u>(approx. 12</u></b> <b><u>lessons)</u></b>	<b><u>Summer HT2</u></b> <b><u>approx. 6</u></b> <b><u>lessons)</u></b>
<ol style="list-style-type: none"><li>1. Elements and compounds</li><li>2. Formulae and equations</li><li>3. Mixtures</li><li>4. History of the atom</li><li>5. Structure of the atom</li><li>6. Ions, atoms and isotopes</li><li>7. Electronic structure</li><li>8. Development of the periodic table</li><li>9. Metals and non metals</li><li>10. Exploring group 0</li><li>11. Exploring group 1</li></ol>	<ol style="list-style-type: none"><li>1. Graphene and fullerenes</li><li>2. Metallic bonding</li><li>3. Properties of metals and alloys</li><li>4. Nanoparticles and applications</li><li>5. Metal oxides</li><li>6. Reactivity series</li><li>7. Displacement</li><li>8. Extraction of metals and reduction</li><li>9. Acids and metals</li><li>10. Neutralisation</li><li>11. Soluble salts</li><li>12. RP making a salt</li></ol>	<ol style="list-style-type: none"><li>1. RP Electrolysis</li><li>2. Electron transfer</li><li>3. Balancing equations</li><li>4. Conservation of mass</li><li>5. Relative formula mass</li><li>6. Moles</li><li>7. Amounts of substances</li><li>8. Moles to balance equations</li><li>9. Concentrations of solutions</li><li>10. Limiting reactants</li></ol>	<ol style="list-style-type: none"><li>1. Volume of gases</li><li>2. Endothermic and exothermic reactions</li><li>3. RP temperature changes</li><li>4. Energy level diagrams</li><li>5. Energy change calculations</li><li>6. Cells and batteries</li><li>7. Fuel cells</li><li>8. Measuring rates</li><li>9. Collision theory and rates</li><li>10. The effect of temperature</li></ol>	<ol style="list-style-type: none"><li>1. Reversible reactions and energy change</li><li>2. Equilibrium</li><li>3. Changing concentration and equilibrium</li><li>4. Changing temperature and equilibrium</li><li>5. Changing pressure and equilibrium</li><li>6. Crude oil and hydrocarbons</li><li>7. Fractional distillation</li><li>8. Properties of hydrocarbons</li><li>9. Combustion</li></ol>	<ol style="list-style-type: none"><li>1. Carboxylic acids</li><li>2. Additional polymerisation</li><li>3. Condensing polymerisation</li><li>4. Amino acids</li><li>5. DNA</li><li>6. Pure substances</li><li>7. Formulations</li><li>8. Chromatography</li><li>9. Tests for gases</li></ol>

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<ul style="list-style-type: none"> <li>12. Exploring group 7</li> <li>13. Reaction trends and predicting reactions</li> <li>14. Transition metals</li> <li>15. States of matter</li> <li>16. Ionic bonding</li> <li>17. Properties of ionic compounds</li> <li>18. Covalent bonding</li> <li>19. Properties of small molecules</li> <li>20. Polymer structures</li> <li>21. Giant covalent structures</li> </ul>	<ul style="list-style-type: none"> <li>13. pH and neutralisation</li> <li>14. Titration</li> <li>15. Strong and weak acids</li> <li>16. The process of electrolysis</li> <li>17. Electrolysis of molten ionic compounds</li> <li>18. Using electrolysis to extract metals</li> <li>19. Electrolysis of aqueous solutions</li> </ul>	<ul style="list-style-type: none"> <li>11. Percentage yield</li> <li>12. Atom economy</li> </ul>	<ul style="list-style-type: none"> <li>11. The effect of concentration and pressure</li> <li>12. The effect of catalysts</li> <li>13. RP rate of reaction</li> </ul>	<ul style="list-style-type: none"> <li>10. Cracking and alkenes</li> <li>11. Structure and formula of alkenes</li> <li>12. Reactions of alkenes</li> <li>13. Alcohols</li> </ul>	
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