

This course will appeal to those students who:

Are willing to think around problems

Have an interest in all of the science subjects

Enjoy the challenge of using scientific knowledge in new situations

Units of Study

Foundations of Chemistry

- Atoms, compounds and molecules.
- Acid-base and redox reactions
- ♦ Moles and equations

Electrons, bonding and structure

Qualitative analysis

Periodic Table and Energy

- The periodic table and periodicity
- Enthalpy change

Physical Chemistry and Transition Elements

- pH and buffers
- Redox and electrode potentials
- Gal Oliginistry and Transition Lightenia
- Core Organic Chemistry
 - Basic concepts
 - Organic synthesi
- Organic Chemistry and Analysis
 - Aromatic compounds
 - Organic synthesis

- Group 2 and the halogens
- Reaction rates and equilibrium
- ♦ Enthalpy, entropy and free energy
 ♦ Transition elements
- Reaction rates and Equilibrium (quantitative)
- Hydrocarbons
- Analytical techniques (IR, MS)
- Alcohols and haloalkanes
- Carbonyls, Carboxylic acids + Esters
- Chromatography + Spectroscopy
- Nitrogen Compounds
- Polymers

Practical Skills in Chemistry

This unit assesses practical and investigative skills developed within contexts encountered during the A Level Chemistry course. This practical skills unit is teacher assessed and externally moderated and involves three tasks; one qualitative, one quantitative and one evaluative.

Entry Requirements

Level 6 or above in GCSE Combined Science, or level 6 or above in GCSE Chemistry. Level 6 in GCSE Mathematics is also required.

What will this course prepare me for?

Chemistry combines well with other scientific disciplines in addition to Maths as an excellent basis for a degree in Medicine, Pharmacy or Pharmacology. The numerical, analytical and problem-solving skills you develop in Chemistry are recognised as very useful for careers in Accountancy and Computing.