

## Personalised Learning Checklists AQA Chemistry Paper 1 Separate Science and Combined Science

AQA Chemistry (8462) from 2016 Topics C4.1 Atomic structure and the periodic table						
Торіс	Student Checklist	R	Α	G		
1.1 A simple model of the atom, symbols, relative atomic mass, electronic charge	Atoms, elements and compounds					
	Mixtures					
ple mc atom, relativ mass, c charg	The development of the model of the atom					
mp e a ls, r iic r	Relative electrical charges of subatomic particles					
L A simple m of the atom, mbols, relati atomic mass ectronic char	Size and mass of atoms					
1.1 A simp of the a symbols, r atomic r electronic	Relative atomic mass (revisit in 3)					
4.1.1 0 sym ai ai	Electronic structure					
e	The Periodic table					
abl	Development of the periodic table					
ict	Metals and non-metals					
iod	Group 0					
The periodic table	Group 1					
	Group 7					
	Properties of transition metals					
4.1.2	Comparison with Group 1 elements					
4	Typical properties					

AQA Chemistry (8462) from 2016 Topics C4.2 Bonding, structure, and the properties of matter					
Торіс	Student Checklist	R	Α	G	
to	The states of matter (revisit in 3 and 8 Chem)				
	State symbols				
	Chemical bonds				
	Ionic bonding				
covalent e are rel tances	lonic compounds				
onds, ionic, coval metallic and structure are rties of substance	Properties of ionic compounds				
ionic, llic tructur of subs	Covalent bonding				
onds, ioi metallic and stru ties of s	Properties of small molecules				
4.2.1 Chemical bonds, meta .2 How bonding and s the properties (	Giant covalent structures				
bo n gaa bert	Diamond				
ical din rop	Graphite				
.1 Chemical b How bonding the prope	Graphene and fullerenes				
the second	Metallic bonding				
2.1 Ho	Properties of metals and alloys / Metals as conductors				
4.2	Sizes of particles and their properties				
4.	Uses of nanoparticles				

			AQA Chemistry (8462) from 2016 Topics C4.3 Quantitative chemistry			
Торіс			Student Checklist	R	Α	G
1 ica	ıre	:s, rva	Conservation of mass and balanced chemical equations			
4.3.1 hemic	- measure	ments, onserva	Mass changes when a reactant or product is a gas			
ĊP 4	ш.	Ú	Relative formula mass			
se of t of ce in n to f pure		ure s	Moles (HT only)			
		Amounts of substances in equations (HT only)				
L3.2 Use amount ubstanc	4.3.2 Use amount substance relation t nasses of p substance	s o stai	Using moles to balance equations (HT only)			
4.3.2 amo substa	ela.	relation to masses of pu substances	Limited reactants (HT only)			
4 9 8	su s	Ê	Concentration of solutions			
and omy cal		g Dns	Using concentrations of solutions in mol/dm <sup>-3</sup> (chemistry, HT only)			
ield and conomy	Sus,	.4 Using entratior	Use of amount of substance in relation to volumes of gases (chemistry, HT only)			
ž ě ž	.3 Yield ar m econon f chemical eactions.	n C	Yield and atom economy of chemical reactions (chemistry only)			
		4.3.4 Using concentratio	Percentage yield			
4.3 ato	- כ	4 COL	Atom economy			



	AQA Chemistry (8462) from 2016 Topics C4.4 Chemical changes			
Торіс	Student Checklist	R	Α	G
' of	Metal oxides			
4.4.1 Reactivity metals	The reactivity series			
4.4 acti	Extraction of metals and reduction			
Re	Oxidation and reduction in terms on electrons (HT only)			
	Reactions of acids with metals			
	Neutralisation of acids and salt production			
cids	Soluble salts			
4.4.2 Reactions of acids	<b>Required practical 1:</b> preparation of a pure, dry sample of a soluble salt from an			
o st	insoluble oxide or carbonate using a Bunsen burner to heat dilute acid and a water			
tion	bath or electric heater to evaporate the solution			
eac	The pH scale and neutralisation			
2 K	Titrations			
4.4	Chem ONLY: <b>Required practical 2:</b> determination of the reacting volumes of			
-	solutions of a strong acid and a strong alkali by titration			
	Strong and weak acids (HT only)			
	The process of electrolysis			
/sis	Electrolysis of molten ionic compounds			
rol	Using electrolysis to extract metals			
Electrolysis	Electrolysis of aqueous solutions			
Э Э	<b>Required practical 3:</b> investigate what happens when aqueous solutions are			
4.4.3	electrolysed using inert electrodes			
	Representation of reactions at electrodes as half equations (HT only)			

AQA Chemistry (8462) from 2016 Topics C4.5 Energy changes						
Торіс	Student Checklist	R	Α	G		
p su	Energy transfer during exothermic and endothermic reactions					
4.5.1 Exothermic and endothermic reactions	<b>Required practical 4</b> : investigate the variables that affect temperature changes in reacting solutions					
4.5.1 Exothe endothermic	Reaction profiles					
4.5 end	The energy change of reactions (HT only)					
Chemical and fuel cells	Cells and batteries					
4.5.2 Cher cells and cells	Fuel cells					