

The Orme Academy: Science Year 10

Your journey continues onto Year 11 Science

$$\% \text{ ATOM ECONOMY} = \frac{\text{Mr OF DESIRED PRODUCT}}{\text{Mr OF TOTAL PRODUCTS}} \times 100$$

Fuel cells

Cells and batteries

Investigating temperature change

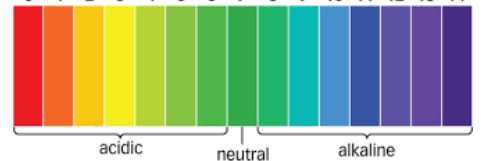
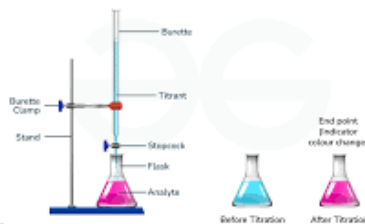
Percentage yield

Atom economy

Exothermic and endothermic

Bond energy (HT only)

C4.4 Energy changes



Volumes of gases

Strong and weak acids (HT only)

Titration

Acids, alkalis and neutralisation

Acid + Alkali → Salt + Water

nitric acid + potassium hydroxide → potassium nitrate + Water

Limiting reactants (HT only)

Concentration

C4.3 Quantitative Chemistry

Obtaining raw materials

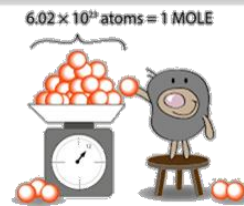
Recycling metals



Corrosion

Electrolysis RP

Moles (HT only)



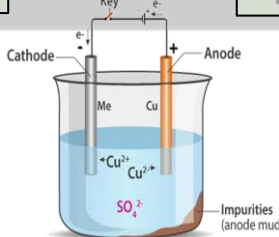
Extracting metals by Electrolysis

Electrolysis

Half equations

Displacement Reactions

Ionic equations (HT only)



most reactive
K
Na
Ca
Mg
Al
C
Zn
Fe
Sn
Pb
H
Cu
Ag
Au
Pt
least reactive



Extracting metals

C4.2 Extraction of metals

Polymers

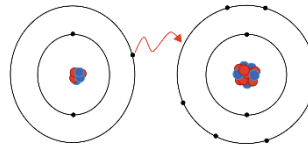
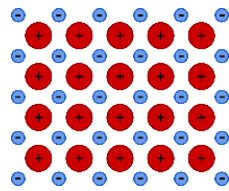
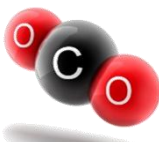
Metallic Bonding

Alloys

Bonding models

Nanoparticles

Allotropes of Carbon



Your journey starts here!

Covalent substances

Covalent Bonding

Ionic substances

Ionic Bonding

C4.1 Structure and Bonding